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New 1,4-diazepine derivatives having anti-ulcer activity.

The invention relates to a group of new 1,4-diazepine derivatives of the formula

$$A \longrightarrow R_3$$

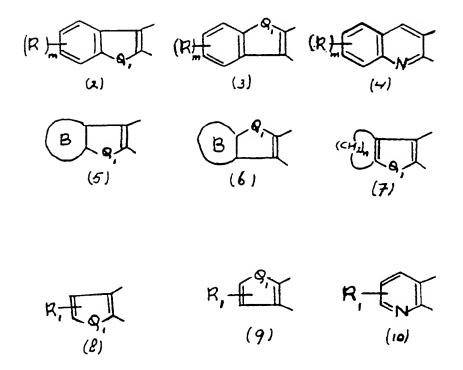
$$R_3$$

$$R_4$$

$$R_5$$

wherein A represents a group of the formulae 2-10

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These compounds have a strong anti-ulcer activity after oral administration.

New 1,4-diazepine derivatives having anti-ulcer activity

The invention relates to a group of new 1,4-diazepine derivatives and salts and prodrugs thereof having favourable properties on ulcers in the gastrointestinal tract, to compositions which comprise these compounds as an active substance, and to the preparation of the said compounds.

Ulcerations of the stomach and the duodenum are a frequently occurring syndrome in human beings. The objects of pharmaco-therapeutic treatment of these disorders are: relieving pain, curing the ulcer, and preventing recurrence of the symptoms.

Present-day pharmaco-therapy is mainly directed to inhibiting the secretion of one of the agressive factors in stomach and duodenum, namely the gastric acid. The histamine H₂-antagonists like cimetidine and ranitidine are the best known examples hereof. Besides, a few pharmaco-therapeutic substances are known which have so-called mucosa-protective properties. This means that the said substances, in oral dosages which do not inhibit gastric acid secretion, favourably influence one or more of the factors which contribute to a correct balance between agressive effects (for example, gastric acid, pepsin, bile acids) and defensive effects (for example, mucus secretion, bicarbonate secretion, blood circulation) on the mucosa of stomach and duodenum. The best known examples hereof are sucralphate and bismuth subcitrate.

It is the object of the present invention to provide compounds which have a combination of the above-described properties i.e.

- cause inhibition of gastric acid secretion, for example by a reversible inhibition of the (H + K)-ATPase system in the stomach.
- have a so-called mucosa-protective effect.

Both effects should occur after oral administration and should continue for a sufficiently long period of time.

On the basis of their properties such compounds could also be used for one or more of the following disorders: gastritis, duodenitis, reflux oesophagitis, non-ulcer dyspepsia and the Zollinger-Ellison syndrome.

It has now been found surprisingly that the compounds of formula 1 hereinafter and their acid addition salts satisfy the objects mentioned hereinbefore.

$$\begin{array}{c|c}
R_1 & R_3 \\
\hline
R_5 & R_4
\end{array}$$
(1)

In formula 1 the symbols used have the following meanings:

-A together with the two carbon atoms of the seven membered ring, forms a group of the formulae 2-10

$$(R)$$
 (R) (R) (R) (R) (A) (A)

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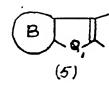
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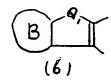
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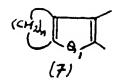
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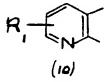
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R, (9)



whereir

R is halogen, alkyl, alkoxy, alkylthio, amino, mono- or dialkylamino, hydroxyalkyl, alkylcarbonyl, aminocarbonyl, mono- or dialkylamino carbonyl, alkoxycarbonyl, nitro, cyano, trifluoromethyl, trifluoromethoxy, aikylsulphonyl, aminosulphonyl, hydroxy, alkylene dioxy, phenyl or benzoyl, and m has the value 0-4, Q₁ is oxygen or sulphur,

B together with the two carbon atoms of the five membered ring, is thienyl or pyridyl, which groups may be substituted with a group $(R)_m$,

R₁ is a phenyl group optionally substituted with group (R)_m,

n has the value 3 or 4:

R₂ is hydrogen, alkyl, phenylalkyl, alkylcarbonyl, aminocarbonyl, mono- or dialkylaminocarbonyl, alkoxycarbonyl or alkoxycarbonylalkyl:

 $\mbox{\bf R}_{3}$ and $\mbox{\bf R}_{4}$ independently of each other are hydrogen, alkyl or hydroxy; and

 R_5 is phenyl, thienyl, cyclopentyl, cyclohexyl, n-pentyl, n-hexyl, benzyl, phenylethenyl, phenylamino, or benzylamino, which groups may be substituted with a group $(R)_m$.

When in the above formula (1) R_2 is hydrogen, tautomerism may occur in the diazepine ring. As a result of the occurrence of the said tautomerism, an optional group R_2 having a meaning different from hydrogen, may be bonded to the other nitrogen atom of the diazepine ring. In so far as tautomerism is concerned, the invention relates to compounds substituted with a group R_2 which is present either at the nitrogen atom in position 1 or at the nitrogen atom in position 4.

When groups R_3 and/or R_4 in the above formula (1) have a meaning different from hydrogen, the carbon atoms to which R_3 and/or R_4 are bonded are chiral centres. In so far as chiral centres are concerned, the invention relates to the various enantiomers of the compound of formula 1 and to racemic mixtures of the said compounds.

On the basis of their properties the invention preferably relates to compounds of formula 1, wherein R, Q_1 , \underline{m} and R_2 have the above mentioned meanings, and A is a group of the formulae 2, 3, 5 or 6, B is the thienyl group substituted with $(R)_m$, R_3 and R_4 are hydrogen, and R_5 is a phenyl group or thienyl group substituted with a group $(R)_m$.

The activity of the compounds was determined in some relevant test models, and compared with known anti-ulcer substances, i.e. the histamine H_2 -antagonists cimetidine and ranitidine:

1a. Inhibition of gastric acid secretion in vitro

In vitro acid secretion was investigated by studying the uptake of the weak base (14C)-aminopyrine in intact parietal cells from the stomach of the rabbit. Acid secretion was stimulated by dibutyryl-cyclic-AMP and the inhibition of acid secretion was determined by the decrease of the (14C)-aminopyrine accumulation in the parietal cells (according to a modification of the method described by T. Berglindh et al., Akta Physiol. Scand. 97, 401, 1976). The results were expressed in pl₅₀-values.

1b. Inhibition of gastric acid secretion in vivo

In vivo acid secretion was examined by determining the quantity of acid in the stomach of the pylorus-ligated rat (H. Shay et al., Gastroenterology 5, 43, 1945). Acid secretion was stimulated by subcutaneous administration of histamine. The test substances were administered orally and the inhibition of acid secretion was established by comparison of the quantity of gastric acid produced in 1 hour in treated and untreated animals. The results were expressed in ID50-values.

Mucosa-protective activity in vivo

In vivo mucosa-protective activity was investigated by studying the protection against ethanol-induced stomach damage in the rat (A. Robert et al. Gastroenterology 77, 433, 1979). The test substances were administered orally and the mucosa-protective activity was established by comparison of the stomach damage caused by ethanol in 1 hour in treated and untreated animals. The results were expressed in ED₅₀-values.

3. Anti-ulcer activity in vivo

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The arti-ulcer activity for the most active compounds was determined by studying the protection against differently caused damage of the stomach and the duodenum in the rat. Notably the protection was studied against aspirin, indomethacin and stress-induced damage of the stomach and damage of the duodenum induced by cysteamine. The results were expressed in ED₅₀-values.

The compounds according to the invention of formula 1 are new compounds with the exception of the compound in which A is a group of formula 2, wherein \underline{m} is 0, Q_1 is oxygen, R_2 , R_3 and R_4 are hydrogen and R_5 is phenyl. This compound is known from J. Heterocyclic Chem. 20, 1251, (1983).

The compounds may be prepared in a manner known per se for analogous compounds.

Depending on the meanings of the symbols, the compounds of formula 1 can be obtained inter alia by means of one of the following methods.

Compounds of formula 1, wherein A is a group of formula (2) or (5) can be obtained, for example, by converting a compound of formula 11

wherein Q_1 has the above-mentioned meaning and $B^{'}$ is a phenyl ring substituted with $(R)_m$, or a group B having the above-mentioned meaning, with a compound of formula 12

wherein R₅ has the above-mentioned meaning and L is a halogen atom.

The reaction is preferably carried out in an inert solvent, for example, methanol, ethanol, acetonitrile or dimethyl formamide, at a temperature of 0 to 180°C for 1-48 hours. A base, for example, sodium methoxide, may be added to the reaction mixture.

The compounds of formula 11 are partly known compounds (German Patent Specification 2.034.756 and 2.034.987) and, in so far as they are new compounds, they may be prepared in an analogous manner.

Furthermore, the compounds of formula 1, wherein A is a group of formula (2) or (5), can be obtained by converting a compound of formula 13

$$\begin{array}{c}
A' \\
C \\
C \\
B \\
C
\end{array} - R_{5}$$
(13)

wherein A is the group (2) or (5) and R_5 and L have the above-mentioned meanings, with an amine of the general formula 14

 $H_2N-CH-CH-NH_2$ (14)

wherein R₃ and R₄ have the above-mentioned meanings.

The reaction is preferably carried out in an inert solvent, for example, acetonitrile or dimethyl sulphoxide, at a temperature of 20 to 200 °C for 1-72 hours.

Some of the compounds of formula 13 are known (Eur.J.Med.Chem.Chim.Ther.20, 425, 1985). In so far as the compounds are new they can be obtained in analogous manners.

Moreover, compounds of formula 13, wherein Q₁ (in A') is a sulphur atom can also be obtained in a simple manner known per se from the readily available acid chlorides of formula 15 (Synthesis 670, 1981):

wherein B' has the above-mentioned meaning.

Furthermore, the compounds of formula 1, wherein A is a group of the formulae (2)-(10) can be obtained, for example, by converting compounds of formula 16

$$\begin{array}{c}
A \\
C - R_{s}
\end{array}$$
(16)

wherein A and R₅ have the above-mentioned meanings, into the corresponding carbamates of formula 17 in a manner known per se (J.Heterocyclic Chem 20, 1251, 1983):

alkylating these with a compound of formula 18

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$$L - CH - CH - NH - C - OC_2H_5$$

$$R_3 R_4 O$$
(18)

wherein R₃, R₄ and L have the above-mentioned meanings, then deprotecting the two amino functions, after which the desired compounds of the general formula 1 are obtained by ring closure.

The compounds of formula 1 can also be obtained by alkylation of compounds of formula 16 with N-carbo-tert.butyloxy-aziridine, followed by removal of the tert.butyloxycarbonyl group and ring-closure.

Compounds having formula (19)

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wherein A, R₄ and R₅ have the above-mentioned meanings, can be obtained in a manner known per se - (J.Heterocyclic Chem. 16, 189, 1979) or quite analogous thereto starting with compounds of formula 16. The compounds of formula 19 may then be converted into compounds of formula 1 by reduction with, for example, LiAlH₄.

The compounds of formula 16 are partly known compounds (J.Prakt.Chem. 315, 779, 1973 and J.Org.Chem. 39, 3440, 1974) and, in so far as they are new compounds, they can be obtained in analogous manners.

As last reaction step one or more chemical conversions known per se, for example, reduction reactions, acylation reactions, alkylation reactions, and the like, may be used to obtain the desired compounds of formula 1.

As examples of pharmaceutically acceptable acids with which the compounds of formula 1 can form salts may be mentioned hydrochloric acid, sulphuric acid, phosphoric acid, methanesulphonic acid, ptoluene sulphonic acid, benzoic acid, acetic acid, propionic acid, tartaric acid, succinic acid, citric acid, fumaric acid, maleic acid, etc.

The compounds of formula 1 and the salts thereof can be brought into a form preferably suitable for oral administration, for example, capsules, tablets, coated tablets, and pills, by means of conventionally used techniques and auxiliary substances.

The invention will now be described in greater detail with reference to the ensuing specific examples.

EXAMPLE I

5-(4-chlorophenyl)-2.3-dihydro-1H-benzothieno-[3,2-e]-1,4-diazepine hydrobromide

A solution of 2-(2-mercaptophenyl)-2-imidazoline (17.8 g; 0.1 mol) and 2-bromo-4 -chloroacetophenone (23.4 g; 0.1 mol) in a mixture of acetonitrile (150 ml) and methanol (50 ml) was heated at 60 °C while stirring for 3 hours.

The reaction mixture was then evaporated under reduced pressure. The residue was dissolved in a mixture of isopropanol (160 ml) and methanol (40 ml) and the reaction mixture was heated at reflux temperature for 5 hours.

The mixture was cooled to room temperature and the precipitate formed was filtered off, washed with isopropanol and liberated from solvent residues. In this manner 32.5 g of 5-(4-chlorophenyl)-2,3-dihydro-1H-benzothieno-[3,2-e]-1,4-diazepine hydrobromide were obtained; melting-point 317-319 °C.

The compounds recorded in the following table A were prepared in an analogous manner:

TABLE A

	Comp.	z_1	z ₂	z_3	z ₄	salt	melting-point
15	no.						(°C)
	1	н	Н	Н	н	HBr	293 - 295
20	2	Н	Н	Н	4-NO ₂	HBr	298 - 300
	3	H	н	Н	4-Br	HBr	285 - 289
	4	H	Н	Н	4-0CH ₃	HBr	263 - 265
25	5	H	н	Н	4 - OH	HC1	328 - 330
25	6	H	н	Н	4-CF ₃	HBr	326 - 327
	7	H	Н	н	3-CF3	free base	1100
	8	H	Н	Н	2-CF ₃	free base	90 - 95
30	9	Н	н	Н	4 - C≣N	HBr	312 - 314
	10	Н	Н	Н	4-COOCH3	HC1	255 - 257
	11	H	н	Н	4-CH2OH	free base	220 - 221
35	12	H	Н	Н	2-0CH ₃	HBr	245 - 248
	13	н	H	н	2 - OH	HBr	340 - 345
	14	Н	H	Н	2 - NO ₂	HC1	> 300
	15	H	н	н	2-C1	free base	242 - 246
40	16	10-F	Н	н	4 - C1	HBr	290 - 293
	17	9-Cl	Н	Н	4 - NO ₂	HBr	190 - 200
	18	9-C1	Н	Н	4 - F	HBr	> 300

	Comp.	z_1	z_2	z_3	Z ₄	salt	melting-point
	no.						(°C)
5	19	Н	Н	CH ₃	4 - F	HBr	270 - 280
	20	н	Н	CH ₃	4-NO ₂	HBr	280 - 295
	21	Н	CH ₃	Н	4 - Cl	free base	157 - 159
iO	2 2	н	CH2COOE	t H	4-C1	HC1	214
	2 3		/N	<i>i</i>		free base	162 - 164
				>			
:5		~	13/	-N-CH,C	COC,H.		
				ì			
				"			
20	24		H C	2		free base	187 - 188
			_й <u> </u>				
		10 1/V	FN				
25			5				
23			EH;				150 160
	25		,			free base	150 - 160
		15-	برين م				
30		احلاه	_~\ <i>\</i>				
	26		Ver,			free base	220 - 225
35	20		4			2.00	
		~ · · · ·)=N'	-			
40		п	C=CH-(
	27	<u></u>	\%_			free base	218 - 220
		Lank	لرز_ بالأد				
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	1 - 6	lecomp	position	ı			

EXAMPLE II

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55 5-(4-chlorophenyl)-2,3-dihydro-1H-benzofuro-[3,2-e]-1,4-diazepine

A suspension of 2-(2-hydroxyphenyl)-2-imidazoline (16.2 g; 0.1 mol) and 2-bromo-4'-chloroacetophenone (23.4 g; 0.1 mol) in acetonitrile (180 ml) was heated at 50 °C while stirring for 3 hours.

After cooling to room temperature ether (180 ml) was added. The precipitate formed was filtered off and washed with ether (200 ml).

Sodium methoxide (5.4 g; 0.1 mol) was added to a suspension of the resulting product in dimethyl formamide (100 ml) and the mixture was heated at 60 °C while stirring for 4 hours.

A mixture of methanol (35 ml), water (35 ml) and 2N sodium hydroxide (35 ml) was then added. The mixture was cooled to 10 °C and the precipitate formed was filtered off, washed with water and dried.

To a suspension of the product (23.7 g; 0.08 mol) in a mixture of methanol (100 ml) and ether (200 ml) a solution of methanesulphonic acid (7.7 g; 0.08 mol) in ether (50 ml) was added. After addition of petroleum ether (100 ml) the mixture was stirred at 0-5 °C for 30 minutes. The resulting precipitate was filtered off, washed with ether and dried.

In this manner 27.7g of 5-(4-chlorophenyl)-2,3-dihydro-1H-benzofuro-[3,2-e]-1,4-diazepine mesylate were obtained; melting-point 226-227 °C.

The compounds mentioned in Tables B and C hereinafter were obtained in an analogous manner.

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TABLE B

	Соп	\mathbf{p} . \mathbf{z}_1	z ₂	z ₃	salt	melti	ing-point
15	no.					(°C)	
					-		· · · · · · · · · · · · · · · · · · ·
20	28	Н	4-0CF ₃	н	free	base ·	173-174
20	29	H	4-Br	н	HC1		280-281
	30	Н	2 - F	н	free	base	210-211
	31	H	4-CF3	Н	HC1		303-307
25	32	Н	2,4-di-F	Н	HC1		256-258
	33	Н	3,4-di-Cl	н	HBr		294-296
	34	8-0CH ₃	4-C1	н	HC1		200
30	35	9-Br	4 - F	Н	free	base	130 [©]
	36	7 - F	4 - C1	н	HC1		205
	37	8 - F	4-C1	н	free 1	base	145-147
	38	7 - C1	4 - F	Н	free 1	base	179-181
35	39	8 - Cl	4 - F	н	free 1	base	130
	40	9-0CH ₃	4-C1	н	free 1	base	181-183
	41	9 - OH	4-C1	H	HBr		203-205
40	42	Н	4-C1	CH3	free 1	base	109-111
	43	Н	4-C1	benzyl	HCl		230-231
	44		H		нс1		260
45			N.				
		>> \((J=N				
50				•			

5	Comp. no.	salt melt	ing-point
10	45 John	free base	100-105
15	2н; 46 О П П П П П П П П П П П П П П П П П П	free base	110-120
20	47 0 N N CH - CH - F	HBr	210
30	48 N CH=CH-(-)-1=	free base	170-180
35	49	нсі	218-219
40	$(CH^{\times})^2 - CH^3$	·	

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O-decomposition

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TABLE C

15	Comp.	z _l	z ₂	z ₃ :	salt	melting-point (°C)
.5	50	н	4 - C1	CH ₃	free base	176-178
20	51	н	4 - C1	сосн3	free base	179-181
	5 2 5 3	н н	4-C1 4-C1	СООС ₂ Н ₅ С ₂ Н ₅	free base HCl	132-133 255-256
	54	н	4-C1	benzyl	HC1	1500

1- decomposition

EXAMPLE III

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8-fluoro-5-(4-chlorophenyl)-2,3-dihydro-1H-benzothieno-[3,2-e]-1,4-diazepine

Ethylene diamine (12 g; 0.2 mol) was added to a solution of 6-fluoro-3-chloro-2-(4-chlorobenzoyl)-benzo[b]thiophene (32.5 g; 0.1 mol) (obtained by conversion of 6-fluoro-3-chlorobenzo[b]thiophene carbox-ychloride into the corresponding methoxymethyl amide and then into the desired ketone by means of 4-chlorophenyl-magnesium bromide (see Tet.Letters 22, 3815, 1981)) and the reaction mixture was heated at 60°C while stirring for 24 hours. After cooling to room temperature, ether (100 ml) and 2N hydrochloric acid (250 ml) were added and the reaction mixture was stirred at room temperature for 3 hours.

The ether layer was separated and the acidic water layer was washed once more with ether. (From the collected ether layers 22.7 g (0.07 mol) of the starting substance were obtained by evaporation). The resulting water layer was made alkaline with 2N sodium hydroxide (250 ml) and the precipitate formed was filtered off.

After recrystallisation from acetonitrile 5.9 g of 8-fluoro-5-(4-chlorophenyl)-2,3-dihydro-1H-benzothieno-[3,2-e]-1,4-diazepine were obtained; melting-point 198-199 °C.

The compounds mentioned in Table D hereinafter were obtained in an analogous manner.

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$$Z_{1}$$
 Q_{1}
 Z_{2}
 Q_{3}
 Z_{4}
 Z_{5}

TABLE D

15	Comp.	Q ₁	z ₁	22	salt	melting-point
	5 5	s	10-C1	4-C1	free base	172-173
	5 6	S	Н	4 - F	free base	250-251
20	5 7	S	H	3-C1	free base	198-200
	5 8	S	9 - F	4-C1	free base	192-193
	5 9	S	8-осн ₃	4-C1	free base	164-166
25	60	S	8-CF ₃	4-C1	free base	215
	61	S	8-CH ₃	4-C1	free base	225

5	Comp.	Q ₁	z ₁	z ₂	salt		melting-point (°C)
•	62	S	8 - C = N	4-C1	HC1		282-284
	63	S	7 - F	4 - C1	free	base	9 5 ©
10	64	S	8 - F	4-CF ₃	free	base	109
	6 5	S	9 - F	4-CF3	HC1		296-300
	66	S	9 - F	4 - F	HC1		>300
	67	S	10-CH ₃	4 - Cl	free	base	161-163
15	68	S	9-C1	4 - C1	HC1		>300
	69	S	7 - Gl	4 - C1	HC1		>300
	70	S	9-CF ₃	4 - C1	free	base	132-142
20	71	S	Н	4-N(CH ₃) ₂	free	base	1500
	7 2	S	8,9-OCH ₂ O	4-C1	free	base	1901
	7 3	S	9 - F	4-0CH ₃	HC1		289-291
	74	0	9-NO ₂	4-C1	HC1		>300
25	7 5	0	9-NH ₂	4-C1	нсі		>300
	76	0	9 - C = N	4 - C1	HC1		>300
	77		Н		HC1		150 [©]
30	í	آون ک) } }				
35	78		ñ Fr		HCl		>300
40		[
45	79	J. 6.11			free	base	130-140
50			Y.				

	Comp.	Q ₁	z ₁	Z ₂	salt	melting-point (°C)
5	80	(),	М.— Сн.————— F		нс1	260
15	81				free base	259-261
20	82 <u>j</u>				нсі	>300
25). , 2,	Č.			

O-decomposition

EXAMPLE IV

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5-(4-fluorophenyl)-2,3-dihydro-1H-benzofuro-[3,2-e]-1,4-diazepine

Ethyl chloroformate (32.5 g; 0.3 mol) was added dropwise to a solution of 3-amino-2-(4-fluorobenzoyl)-benzo[b]furan (25.5 g; 0.1 mol) in benzene (300 ml) and pyridine (60 ml). After stirring overnight at room temperature the reaction mixture was extracted with 1N hydrochloric acid (200 ml), 1N sodium hydroxide (200 ml), water (100 ml) and brine (50 ml). The organic layer was dried on sodium sulphate and evaporated under reduced pressure.

The resulting product (approx. 29 g) was added to a suspension of 55% sodium hydride (6.5 g; 0.15 mol) in dimethyl formamide (250 ml). After the occurring gas evolution had stopped, 22.7g (0.15 mol) of ethyl N-(2-chloroethyl)carbamate were added dropwise. After stirring at 60°C for 16 hours, the reaction mixture was poured on ice and extracted three times with ethyl acetate (3 x 250 ml). The collected organic layers were extracted with water (100 ml) and with brine (50 ml), dried on sodium sulphate and evaporated under reduced pressure.

A solution of the resulting product (approx. 27 g) in a 45% HBr-acetic acid mixture (150 ml) was heated at 65°C for 4 hours. After cooling the reaction mixture was poured on ice, made alkaline with aqueous ammonia and extracted twice with dichloromethane (2 x 500 ml). The collected organic layers were washed with water (100 ml), dried on sodium sulphate and evaporated under reduced pressure.

A solution of the resulting product (approx. 21 g) and a 50% sodium hydroxide solution (50 ml) in absolute ethanol (500 ml) was heated at reflux temperature for 90 minutes. After cooling the reaction

mixture was poured on ice and extracted three times with dichloromethane (3 x 300 ml). The collected organic layers were washed with water (100 ml), dried on sodium sulphate and evaporated under reduced pressure. The resulting crude product was purified by means of flash chromatography over 1500 g of silicagel using a mixture of dichloromethane, methanol and ammonia in the ratio 87: 12.5: 0.5 as an eluent. After evaporating the collected fractions under reduced pressure, 10.0 g of 5-(4-fluorophenyl)-2,3-dihydro-1H-benzofuro-[3,2-e]-1,4-diazepine were obtained; melting-point 144-148 °C. The compound no. 83 of the formula

was obtained in an analogous manner as a free base having a melting-point of 147-149 °C.

EXAMPLE V

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5-(4-tr:fluorcmethyl)-6-phenyl-2.3-dihydro-1H-thieno-[2,3e]-1,4-diazepine

A mixture of 2-amino-3-(4-trifluoromethylbenzoyl)-4-phenylthiophene (6.9 g; 20 mmol) and N-carbotent.butyloxy-aziridine (11 mi) and p-toluenesulphonic acid (70 mg) was heated at 130 °C for 14 hours. After cooling other (150 ml) was added to the reaction mixture. The solution was washed with sodium bicarbonate solution (50 ml, 5%) and with brine (50 ml). The organic layer was dried on sodium sulphate and evaporated under reduced pressure. The resulting crude product was purified chromatographically by means of flash chromatography over 200 g of silicagel using a mixture of ether and petroleum ether in the ratio 1:1 as an eluent. After evaporating the collected fractions under reduced pressure, 2.6 g of pure product were obtained.

A solution of the obtained product (2.6 g; 5.4 mmol) in 3N hydrochloric acid (60 ml) was heated at 100°C for 90 minutes. The reaction mixture was poured on ice, made alkaline with ammonia and extracted three times with ethyl acetate (3 x 100 ml). The organic layer was dried on sodium sulphate and evaporated under reduced pressure. The resulting crude product was purified chromatographically by means of flash chromatography over 100 g of silicagel using dichloromethane/methanol/ammonia (93/6.5/0.5) as an eluent. After evaporating the collected fractions under reduced pressure, 0.54 g of pure product were obtained.

A solution of the obtained pure product (0.54 g;1.4 mmol) in pyridine (10 ml), benzene (15 ml) and acetic acid (0.085 g; 1.4 mmol) was heated at reflux temperature for 24 hours while water was separated by means of a Dean-Stark-trap. After cooling the reaction mixture was evaporated under reduced pressure. The resulting crude product was purified by means of flash chromatography over 80 g of silicagel using dichloromethane methanol/ammonia (95/4.5/0.5) as an eluent. After evaporating the collected fractions under reduced pressure 0.23 g of crystalline 5-(4-trifluoromethyl)-6-phenyl-2,3-dihydro-1H-thieno[2,3-e]-1,4-diazepine were obtained; melting point 124-124.5° C.

EXAMPLE VI

5-(4-chlorophenyl)-2.3-dihydro-1H-benzothieno-[2,3-eJ-1,4-diazepine

A solution of 2-amino-3-(4-chlorophenyl)benzo[b]thiophene (28.8 g; 0.1 mol) (Chem.Ber. 101, 1933, 1968) and chloroacetyl chloride (24.8 g; 0.22 mol) in chloroform (350 ml) was heated at reflux temperature for 1 hour and then evaporated under reduced pressure.

A solution of the resulting product (approx. 35 g) and Nal (16.5 g; 0.11 mol) in acetone (500 ml) was heated at reflux temperature for 90 minutes. After cooling to room temperature the NaCl formed was filtered off and the filtrate was evaporated under reduced pressure.

Ammonia gas was led through a solution of the resulting product (approx. 44 g) in a mixture of chloroform (450 ml) and methanol (50 ml) for 3 hours. After stirring at room temperature for another 90 minutes the reaction mixture was extracted with ice-water (250 ml), with a sodium bicarbonate solution (100 ml, 5%), and with brine (100 ml). The organic layer was dried on sodium sulphate and evaporated under reduced pressure.

A solution of the resulting product (approx. 33 g) and acetic acid (6 g; 0.1 mol) in a mixture of pyridine (180 ml) and benzene (200 ml) was heated at reflux temperature for 2.5 hours while water was separated by means of a Dean-Stark- trap. The reaction mixture was evaporated under reduced pressure. The residue was dissolved in dichloromethane (500 ml) and the solution was extracted with a sodium bicarbonate solution (200 ml, 2.5%), with water (200 ml) and with brine (100 ml). The organic layer was dried on sodium sulphate and evaporated under reduced pressure. The resulting crude product was purified chromatographically by means of flash chromatography over 1500 g of silicagel using a mixture of dichloromethane and acetone in the ratio 9:1 as an eluent. After evaporating the collected fractions under reduced pressure, 13.3 g of pure product were obtained.

A solution of the resulting product (13.3 g; 0.04 mol) in tetrahydrofuran (200 ml) was added to a suspension of 7.6 g of lithium aluminium hydride (0.2 mol) in tetrahydrofuran (500 ml) The reaction mixture was stirred at room temperature for 1 hour and then at 50 °C for 30 minutes. After cooling, 7.6 ml of water in 50 ml of tetrahydrofuran, 15.2 ml of 2N sodium hydroxide and 15.2 ml of water, successively, were added dropwise. After stirring at room temperature for a few hours the precipitate formed was filtered off and washed with tetrahydrofuran. The filtrate was evaporated under reduced pressure. The resulting crude product was purified by means of flash chromatography over 1000 g of silicagel using a mixture of dichloromethane, methanol and ammonia in the ratio 90.5: 9: 0.5 as an eluent. After evaporating the collected fractions under reduced pressure, 8.0 g of crystalline 5-(4-chlorophenyl)-2,3-dihydro-1H-benzoth ieno-[2,3-e]-1,4-diazepine were obtained the crystal lattice of which comprised 1 mol equivalent of tetrahydrofuran; melting-point 260 °C.

The compounds mentioned in table E were prepared in a similar manner:

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TABLE E

35	Comp. no.	Structure	Salt	Melting point (°C)
40	84		free base	132

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free base 1700

91 free base 1200

92 HBr >300

Claims

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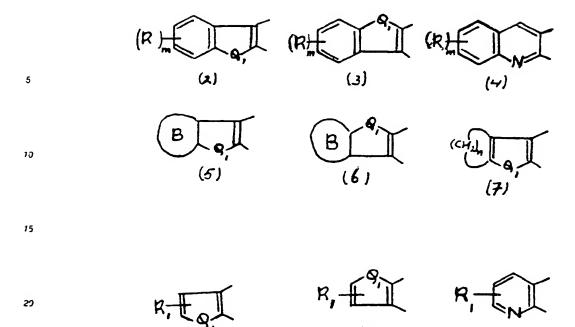
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1. Compositions having anti-ulcer activity which comprise at least one compound of formula 1 as an active substance

 $\begin{array}{c|c}
R_{2} & R_{3} \\
\hline
R_{5} & R_{4}
\end{array}$ (1)

wherein the symbols have the following meanings:

-A together with the two carbon atoms of the seven membered ring, forms a group of the formulae 2-10



wherein

R is halogen, alkyl, alkoxy, alkylthio, amino, mono- or dialkylamino, hydroxyalkyl, alkylcarbonyl, aminocar bonyl, mono- or dialkylamino carbonyl, alkoxycarbonyl, nitro, cyano, trifluoromethyl, trifluoromethoxy, alkylsulphonyl, aminosulphonyl, hydroxy, alkylene dioxy, phenyl or benzoyl, and m has the value 0-4,

Q₁ is oxygen or sulphur,

B together with the two carbon atoms of the five membered ring, is thienyl or pyridyl, which groups may be substituted with a group $(R)_{m_1}$

R₁ is a phenyl group optionally substituted with group (R)_m,

n has the value 3 or 4;

R₂ is hydrogen, alkyl, phenylalkyl, alkylcarbonyl, aminocarbonyl, mono- or dialkylaminocarbonyl, alkoxycarbonyl or alkoxycarbonylalkyl;

R; and R, independently of each other are hydrogen, alkyl or hydroxy; and

 R_5 is phenyl, thienyl, cyclopentyl, cyclohexyl, n-pentyl, n-hexyl, benzyl, phenylamino, or benzylamino, which groups may be substituted with a group $(R)_m$.

or a salt thereof with a pharmacologically acceptable acid.

- 2. Compositions as claimed in Claim 1, characterised in that a compound of formula 1 is used as an active substance, wherein R, Q_1 , \underline{m} and R_2 have the meanings mentioned in Claim 1, A is a group of formula 2, 3, 5 or 6, B is the thienyl group substituted with $(R)_m$, R_3 and R_4 are hydrogen, and R_5 is the phenyl group or thienyl group substituted with $(R)_m$, and salts thereof with pharmacologically acceptable acids
- 3. A method of preparing compositions as claimed in Claim 1, characterised in that a compound of formula 1, wherein the symbols have the meanings mentioned in Claim 1, is brought into a form suitable for administration.
- 4. Compounds of formula 1, wherein the symbols have the meanings mentioned in Claim 1, with the exception of the compound in which A is a group of formula 2, \underline{m} has the value 0, Q_1 is oxygen, R_2 , R_3 and R_4 are hydrogen, and R_5 is phenyl.
 - 5. Compounds as claimed in Claim 4, characterised in that R, Q_1 , \underline{m} and R_2 have the meanings mentioned in Claim 1, A is a group of formula 2, 3, 5 or 6, B is the thienyl group substituted with $(R)_m$, R_3 and R_4 are hydrogen atoms, and R_5 is a phenyl group or thienyl group substituted with $(R)_m$, with the exception of the compound disclaimed in Claim 4.
 - 6. A method of preparing compounds having anti-ulcer activity, characterised in that compounds as claimed in Claim 4 are prepared in a manner known for the synthesis of analogous compounds.
 - 7. A method as claimed in Claim 6, characterised in that compounds of formula 1 are prepared, wherein

A is a group of formula 2 or 5 by converting a compound of formula 11

wherein Q_1 has the meaning mentioned in Claim 1 and B' is a phenyl group substituted with $(R)_m$ or is a group B with the meaning mentioned in Claim 1, with a compound of formula 12

wherein R₅ has the meaning given in Claim 1 and L is a halogen atom.

8. A method as claimed in claim 6, characterised in that compounds of formula 1 are prepared, wherein A is a group of formula 2 or 5, by converting a compound of formula 13

$$\begin{array}{c} A' \\ C \\ -R_5 \end{array}$$

wherein A' is a group of formula 2 or 5, and R₅ and L have the meanings mentioned in Claim 7, with an amine of formula 14

$$H_{2}N-CH-CH-NH_{2}$$
(14)

wherein R₃ and R₄ have the meanings mentioned in Claim 1.

9. A method as claimed in Claim 6, characterised in that a compound of formula 16

$$\begin{array}{c}
A \\
C - R_{5}
\end{array}$$
(16)

when A and R₅ have the meanings mentioned in Claim 1, is converted into the corresponding carbamate of remula 17

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$$0 = C - OC_2H_5$$

$$NH$$

$$C - R_5$$

$$(17)$$

and the carbamate is alkylated with a compound of formula 18

(18)

wherein R₃ and R₄ have the meanings mentioned in Claim 1 and L is halogen.

- 10. A method as claimed in claim 6, characterised in that a compound of formula 16, wherein A and R₅ have the meaning given in claim 1, is reacted with N-carbo-tert.butyloxy-aziridine, followed by removal of the tert.butyloxycarbonyl group en ring-closure.
 - 11. A method as claimed in Claim 6, characterised in that a compound of formula 19

 $\begin{array}{c}
A \\
R_{s}
\end{array}$ (19)

wherein A, R4 and R5 have the meanings mentioned in Claim 1, is reduced.

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11) Publication number:

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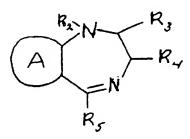
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EUROPEAN PATENT APPLICATION

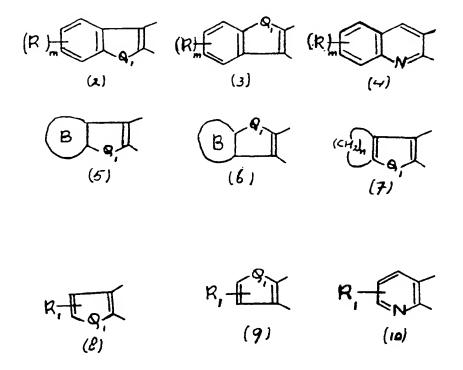
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- New 1,4-diazepine derivatives having anti-ulcer activity.
- The invention relates to a group of new 1,4-diazepine derivatives of the formula



wherein A represents a group of the formulae 2-10



These compounds have a strong anti-ulcer activity after oral administration.

EUROPEAN SEARCH REPORT

EP 89 20 1772

Category	Citation of document with i of relevant pa	ndication, where appropriate, assages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 5)
A	US-A-3 910 887 (DE SILBER) * Claim 1; column 1	UTSCHE GOLD- UND	1	C 07 D 495/04 C 07 D 495/14 C 07 D 491/048 C 07 D 491/147 A 61 K 31/55 (C 07 D 495/04 C 07 D 333:00 C 07 D 243:00) (C 07 D 495/14 C 07 D 333:00 C 07 D 243:00) (C 07 D 491/048 C 07 D 307:00 C 07 D 243:00) (C 07 D 495/14 C 07 D 333:00 -/- TECHNICAL FIELDS SEARCHED (Int. CL5) C 07 D 495/00 C 07 D 491/00 A 61 K 31/00
	The present search report has b	een drawn up for all claims		
	Place of search	Date of completion of the search		Examiner
THE	E HAGUE	15-04-1991	ALF	ARO I.
X:par Y:par doc A:tec	CATEGORY OF CITED DOCUME ticularly relevant if taken alone ticularly relevant if combined with an ament of the same category hnological background newriten disclosure	E : earlier patent do after the filing d other D : document cited i L : document cited f	cament, but publiste ate n the application or other reasons	ished on, or



EUROPEAN SEARCH REPORT

Application Number

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ategory	Citation of document with in of relevant pas	DERED TO BE RELEVAN Idication, where appropriate, ISTAGES	Relevant to claim	CLASSIRICATION OF THE APPLICATION (Int. Cl.5)
				C 07 D 333:00 C 07 D 243:00) (C 07 D 491/147 C 07 D 307:00 C 07 D 243:00 C 07 D 221:00) (C 07 D 495/14 C 07 D 333:00 C 07 D 307:00 C 07 D 243:00)
				TECHNICAL FIELDS SEARCHED (Int. Cl.5)
	The present search report has b	Date of completion of the search		Examiner
X : partic Y : partic	HAGUE LTEGORY OF CITED DOCUME: LULARLY relevant if taken alone LULARLY relevant if combined with and LULARLY relevant of the same category	E : earlier patent de after the filing	ple underlying the ocument, but put date in the application	on or
A: techn O: non-v	ological background vritten disclosure nediate document	&: member of the document	same patent fam	